NUMERICAL INVESTIGATION OF NON-ISOTHERMAL AXIAL SLOSHING OF LIQUID METHANE

N. WEBER, M.E. DREYER

niklas.weber@zarm.uni-bremen.de

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CENTER OF APPLIED SPACE TECHNOLOGY AND MICROGRAVITY



Introduction

- Cryogenics as rocket fuel
 - Phase change and liquid position of importance
- Main engine cut off leads to a step reduction of accelaration
- Axial sloshing might occur
 - Numerical simulations to investigate
 - Validate against experiments
 (Kulev et al., Cryogenics 62 48-59, 2014)







Experimental Setup

- Reorientation of liquid is observed
- 4.7 s of compensated gravity
- Glass cylinder is partially filled with liquid methane
- Pipes and tank filled with gaseous methane
- Pressure, temperature & video are recorded





Experimental Setup

Radius	Fill height	P_0	$T_{sat}\left(P_{0} ight)$
$26.2\mathrm{mm}$	$42\mathrm{mm}$	$47771.5\mathrm{Pa}$	$103.26\mathrm{K}$

- Heaters at top and bottom
 - Bulk liquid heated to saturation condition
 - Axial temperature gradient in wall





Experimental Results

- Reorientation from: high Bo \Rightarrow low Bo
- Damped oscillation of center point
 - Exact absolute position unclear
 - Frequency
 measurable
- Wetting of wall





Experimental Results

Wall temperature decreases after contact with liquid methane







Experimental Results

Increase of vapour pressure





Numerical Tools

- In-house development of OpenFOAM VoF multiphase solver
- Extended by:
 - Weakly compressible treatment of gas phase
 - Phase change model
 - Conjugated heat transfer model



Numerical Setup

- Wedge mesh
- Solid & fluid region
- Conjugated heat transfer
- Linear temperature gradient in vapour and wall
- Fluid at rest
- Rise at wall from prev.
 simulation
- Connecting volume modelled on top of glass















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Wall rise overshoots equilibrium







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Liquid film at wall is formed





time: 0.01 s Numerical Results

 Heat transfer wall from wall to liquid observable

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 Film is influential for thermal condition of wall



Heat transfer from the wall to the fluid gets overestimated







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- Condensation and evaporation occur
- Evaporation at wall
- Condensation in center
 - Rising pressure subcools bulk

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Evaporation outweighs condensation





- Pressure is overestimated
- Evaporation likely overestimated



Phase change most significant influence on pressure





Interface position corresponds to phase change





Conclusion

- Observed phenomena could be reproduced
- Numerical data agrees with experimental data
- Interface position strongly linked to heat and mass transfer in liquid and vapour



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Numerical Tools – Governing Equations

Conservation of mass:

$$\begin{aligned} \frac{\partial \alpha}{\partial t} &+ \frac{\partial \alpha u_j}{\partial x_j} = \frac{S_l}{\rho_l} \\ (1-\alpha) \frac{1}{\beta_2} \frac{\partial p}{\partial t} - (1-\alpha) \frac{\beta_1}{\beta_2} \frac{\partial T}{\partial t} + \frac{\partial \rho_v \left(1-\alpha\right)}{\partial t} + \frac{\partial \rho_v \left(1-\alpha\right) u_j}{\partial x_j} = S_v \end{aligned}$$

- Conservation of momentum: $\frac{\partial \rho u_i}{\partial t} + \frac{\partial \rho u_i u_j}{\partial x_i} = \mu \nabla^2 \mathbf{u} - \frac{\partial p}{\partial x_i} + \rho g_i + f_{\sigma,i}$
- Conservation of energy:

$$\rho\left(\frac{\partial h}{\partial t} + u_j\frac{\partial h}{\partial x_j}\right) = -\left(\nabla\cdot\mathbf{q}\right) + \left(\frac{\partial p}{\partial t} + u_j\frac{\partial p}{\partial x_j}\right) + S_{pc}$$



Gradient based phase change model

Energy balance

 $\dot{m}h_{\rm lv} = q_{\rm l\to int} + q_{\rm v\to int},$

 $\frac{\text{Heat fluxes}}{q_{1 \to \text{int}} = k_1 \nabla_{\text{int},1} T}$ $q_{v \to \text{int}} = k_v \nabla_{\text{int},v} T$

Temperature gradient (interface)

$$\nabla_{\rm int,l} T = \sum_{i} w_i \frac{T_i - T_{\rm sat}}{d_i}$$

$$w_i = \frac{\cos \alpha_i}{\sum_j \cos \alpha_j} \qquad \cos \alpha_i = \frac{(\mathbf{x}_i - \mathbf{x}_{int}) \cdot \mathbf{n}_{int}}{\left| (\mathbf{x}_i - \mathbf{x}_{int}) \right|}$$



Interface temperature:

- dispersion force negligible
- influence of curvature on local sat. cond. negligible

$$\rightarrow T_{\rm int} = T_{\rm sat} \left(p_0 \right)$$



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Conjugate Heat Transfer: Validation





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