



# Liquid Hydrogen Storage: Status and Future Perspectives

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Cryoworld BV



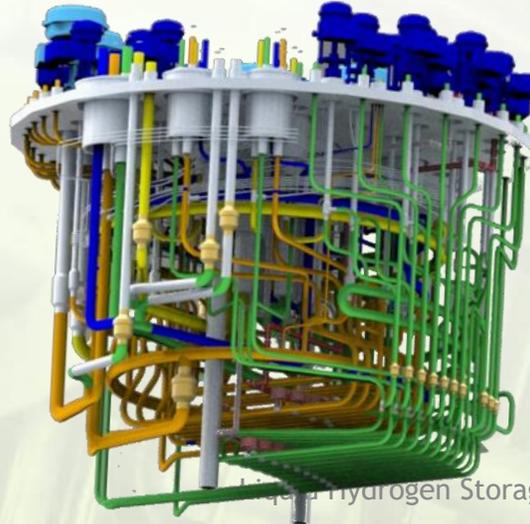
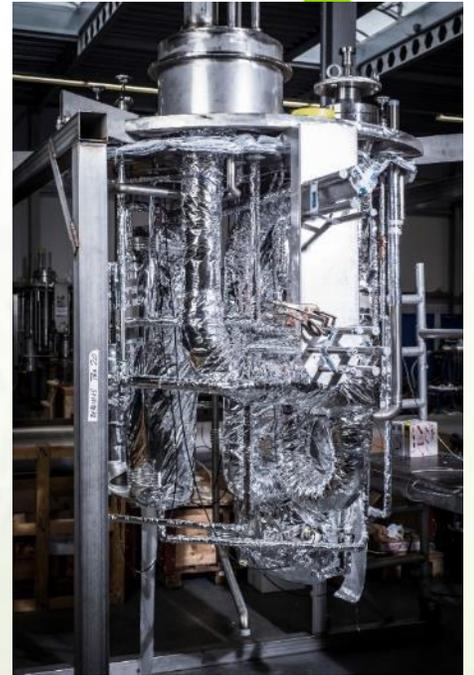
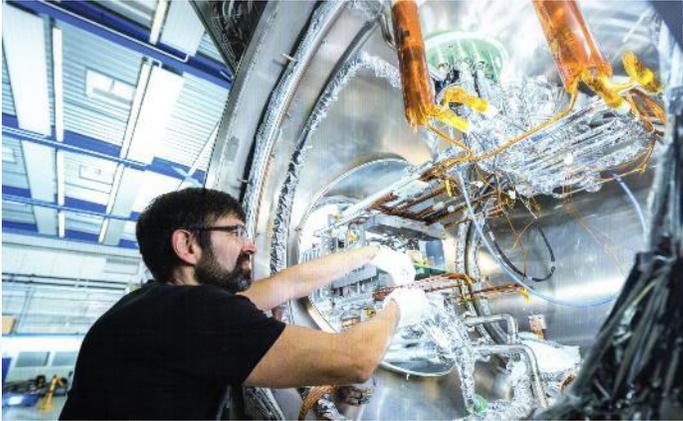
**Cryogenic Heat  
and Mass Transfer**

4<sup>th</sup> - 5<sup>th</sup> November, Enschede, The Netherlands

**CHMT 2019**

# Cryoworld BV

Design, Engineering, Simulation, Manufacturing, Installation and Testing of high-end cryogenic systems for liquid helium, liquid hydrogen and other liquefied gases.



# Outline

- Why hydrogen?
- Why liquid hydrogen?
- Design of liquid hydrogen storage tanks
- Performance of existing liquid hydrogen storage tanks
- Road to high performance liquid hydrogen storage tanks

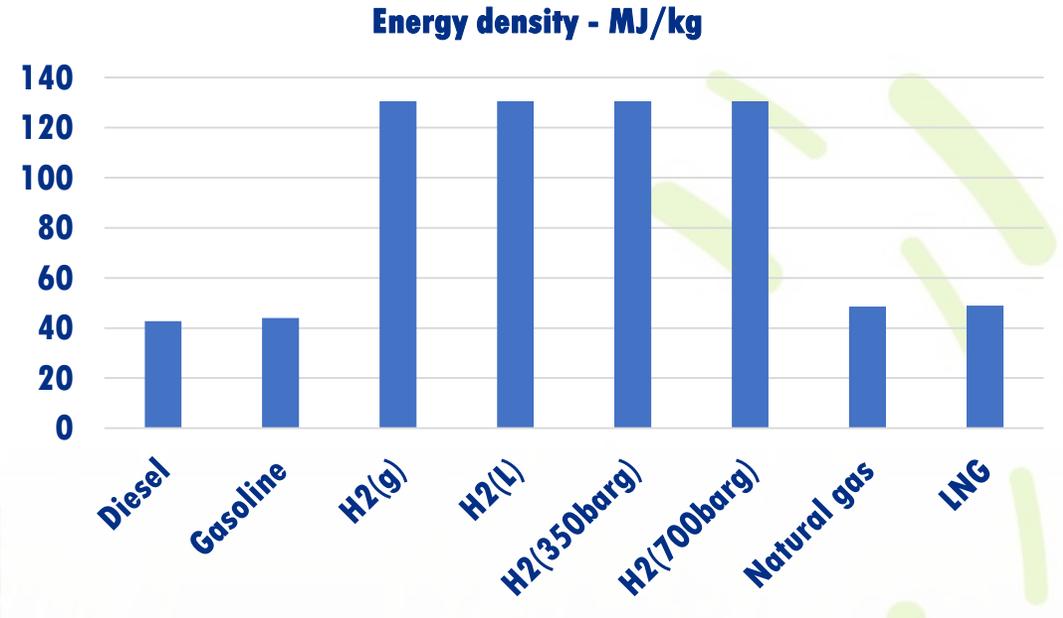


# Why hydrogen?

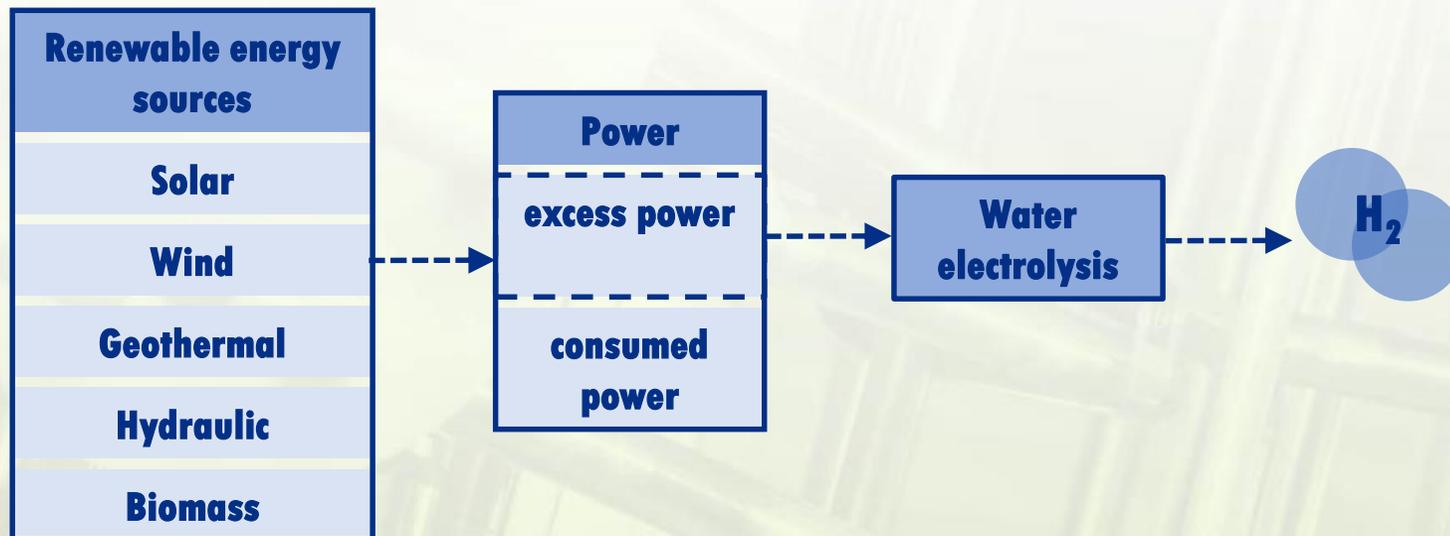
Reducing fossil fuels

Very high energy density

High potential for transportation,  
energy carrier and energy storage



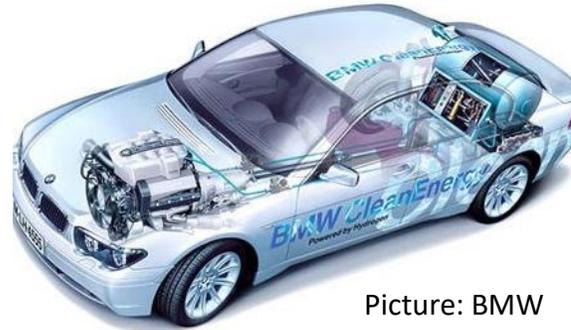
“Green” production when using renewable energy sources



# Hydrogen transportation potential



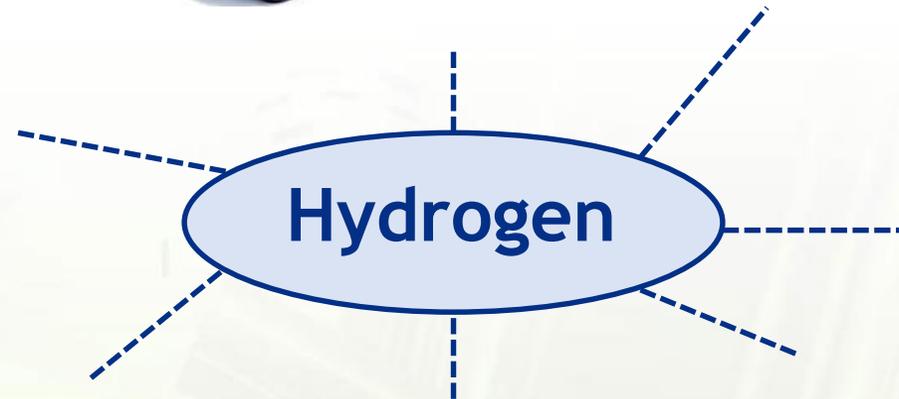
Picture: ESA/CNES/ArianeSpace



Picture: BMW



Picture: Toyota



Picture: Hyundai



Picture: Toyota



Picture: Sandia National Lab



Picture: Airbus

# Hydrogen storage

## Gaseous storage

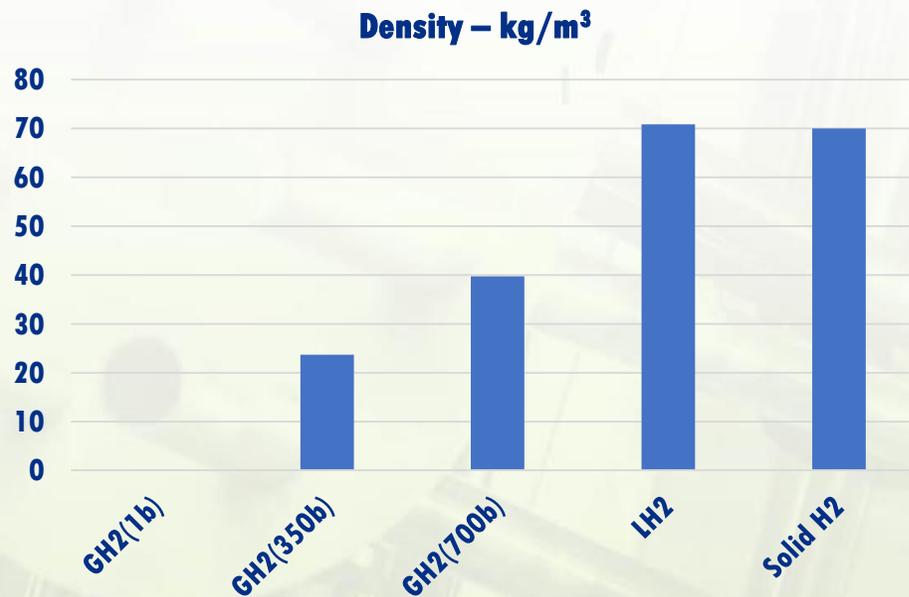
Pressurized to 700-900 bar  
Storage at room temperature

## Liquid storage

At atmospheric pressure  
Storage at 20.4 K  
In slush to increase density

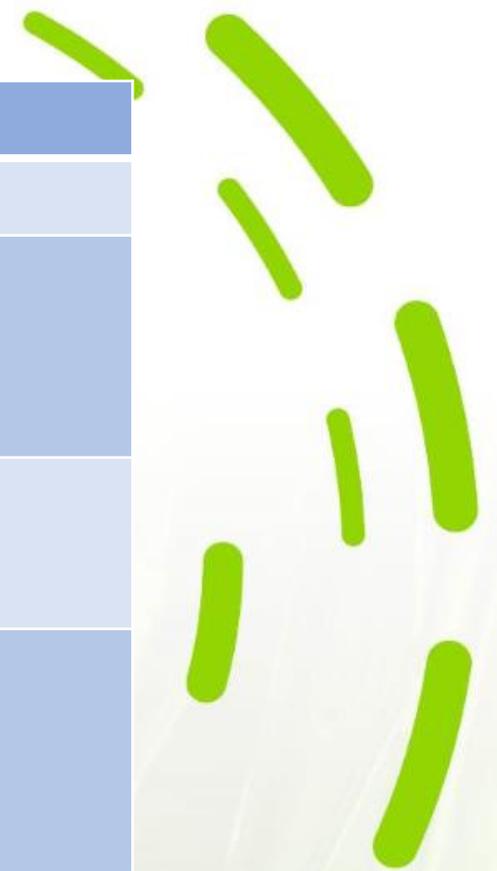
## Solid storage

Physisorption in porous materials  
Adsorbed on metal hydrides  
Complex compounds  
Metals and complexes with water



# Liquid versus high-pressure gas storage

	Liquid storage	High pressure gas storage
Density	70.9 kg/m <sup>3</sup> @ 1 bar, 20.3 K	39 kg/m <sup>3</sup> @ 700 bar, 293 K
Safety	Low pressure system with low enthalpy Spill can lead to floor accumulation Intrinsically safe due to vacuum jacket Boil-off	Huge amount of potential energy Spill can lead to jet No safety barrier
Energy needed	~12 kWh/kg for liquefaction big expensive system boil-off during no use	~ 6 kWh/kg for compression small less expensive system No loss during no use
Handling	Liquid to liquid: by pumping or by gravity Liquid to HP gas: by efficient pumping Very fast filling Simple logistics	Gas to gas: pressure will balance  Slow filling Complex logistics
Storage	Single tank	Long thick walled cylinder bundles
Cost of storage system	€300 / kg Long lifespan	€900 / kg Relatively short lifespan



# Liquid hydrogen storage tanks

## Aviation

Relatively small tank size  
Low weight  
Non-vacuum insulated  
Higher evaporation could be accepted

### Materials

- Aluminium alloys
- Composites
- Fibre reinforced polymers

## Ground-based

Large tanks  
Weight not very important  
Vacuum-insulated  
Zero boil-off



# Liquid hydrogen storage tanks



Picture: NASA

NASA, 3800 m<sup>3</sup>, 270 t  
Boil-off ~12% H<sub>2</sub>



Picture: Kawasaki

JAXA (Kawasaki), 540 m<sup>3</sup>, 38 t



Picture: Linde



Picture: Linde

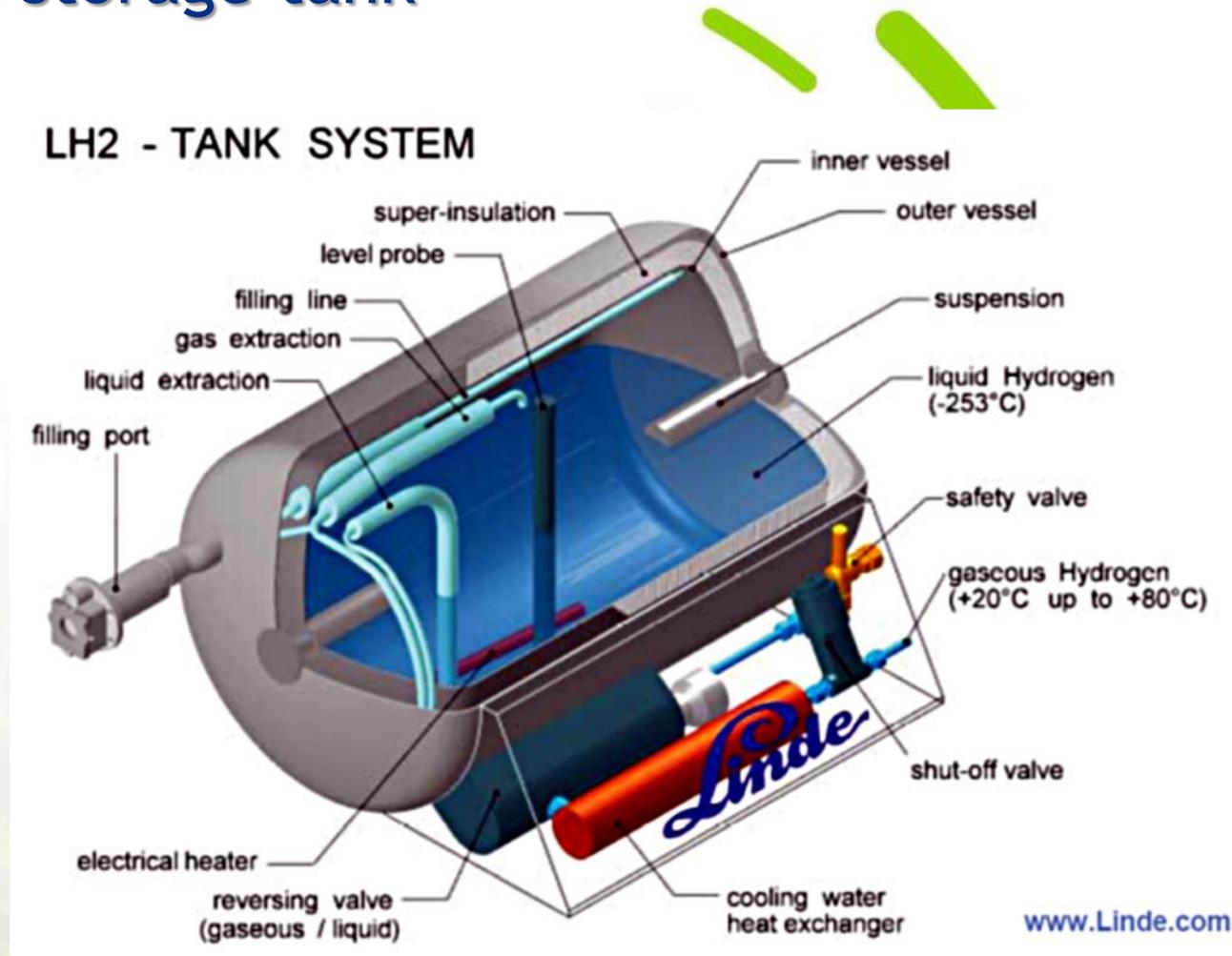
LH<sub>2</sub> truck, < 50 m<sup>3</sup>, < 3.5 t



Largest storage tanks constructed for space applications.  
Spherical shape to optimize surface area to volume ratio.  
Most tanks made with perlite insulation.  
Boil-off rates of 1 - 5% / day.

# Heat flow into liquid hydrogen storage tank

- Insulation: radiation, convection, conduction
- Conduction through support system
  - low thermal conductivity materials
  - high strength materials
- Conduction through interconnecting piping system
  - Small cross-section (thin walled pipes)
  - Increase length
- Radiation from warmer parts of the container
  - Shield by using baffles
  - Optimize design, avoid direct view to warm parts
- Natural convection in vapour above liquid due to heating
- Ortho-para conversion



# Optimizing heat flows

No boiling takes place!

All heat entering the liquid is absorbed primarily by convection.

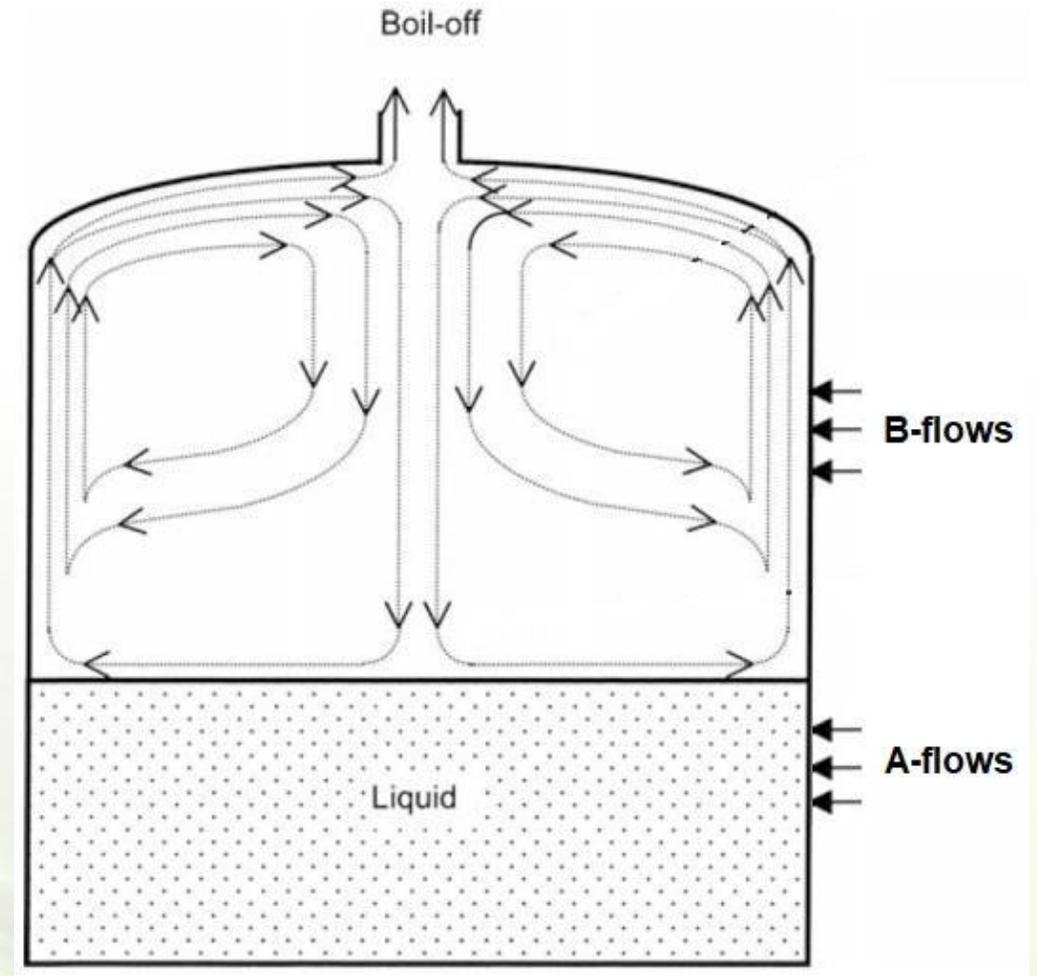
## Two types of heat flows

A: heat flows adsorbed in the liquid resulting in evaporation of the liquid

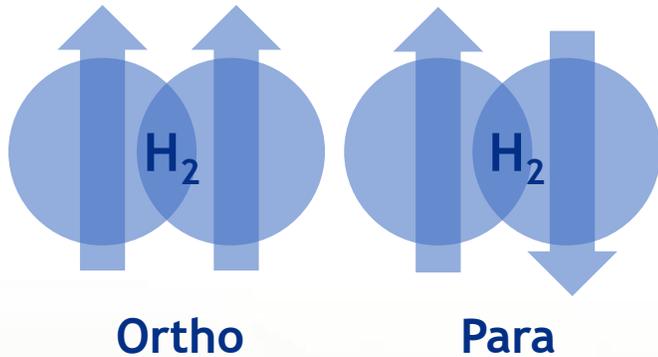
B: heat flows adsorbed in the cold vapour.

With good design, B heat flows may not contribute to the evaporation at all.

Minimize A heat flows and ensuring adsorbing B heat flows in cold vapour.



# Ortho-para conversion



In equilibrium:

25% para-H<sub>2</sub> / 75% ortho-H<sub>2</sub> at RT

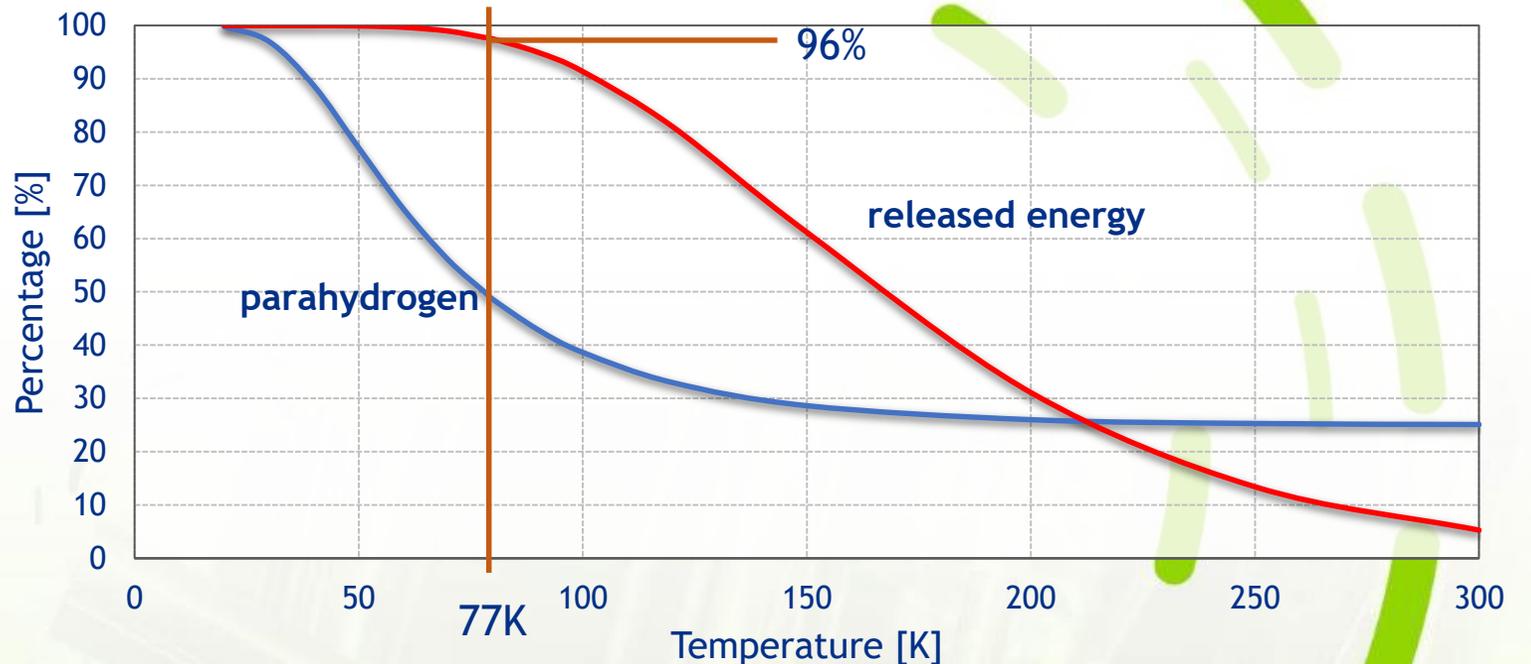
50% para-H<sub>2</sub> / 50% ortho-H<sub>2</sub> at 77 K

99.8% para-H<sub>2</sub> / 0.2% ortho-H<sub>2</sub> at 20 K

Energy released during full conversion at 20 K is ~670 kJ/kg.

A catalyst (i.e. Iron(III) oxide) is used to during liquefaction to speed up the transition.

A small part (~4%) of released energy still will be adsorbed in liquid.



# Insulation

Insulation type	Examples	k-value (300 - 77 K) [mW/mK]	+	-
Insulation@atm pressure	Foams Powders Solid fibres Silica aerogels	~20 - 50	Low weight Relatively cheap Easy to produce	High heat load Need constant purging
Perlite @ $10^{-2}$ mbar		~1.0	Standard technology Good performance	Needs strong vacuum enclosure Heavy structure
Multilayer insulation (MLI) @ $10^{-4}$ mbar		0.0065 - 0.1	Excellent performance	Needs strong vacuum enclosure Heavy structure Most expensive solution

# Current status

Most tanks made with perlite insulation, using LN<sub>2</sub> technology.  
Boil-off rates of 1-5% / day.

## Why LN<sub>2</sub> based storage tanks?

- Experience from the past, standard product;
- Latent heat of H<sub>2</sub> is very high.

Liquid @ 1 bar abs.

	Temperature [K]	Latent heat / mass [kJ/kg]
Hydrogen	20.3	448.9
Nitrogen	77.2	199.3
Methane	111.5	511.1
Helium	4.2	20.8



JAXA (Kawasaki), 540 m<sup>3</sup>, 38 t

# Reducing boil-off

The density of H<sub>2</sub> is very low, resulting in a low volumetric latent heat!

To reduce the boil-off rate it would be better to use LHe technology:

- Multi-layer insulation in combination with high vacuum;
- Actively cooled radiation shields.

Liquid @ 1 bar abs.

	Temperature [K]	Latent heat / mass [kJ/kg]	Density [kg/m <sup>3</sup> ]	Latent heat / volume [kJ/m <sup>3</sup> ]	Factor of LH <sub>2</sub> [-]
Hydrogen	20.3	448.9	70.9	31828	1.0
Nitrogen	77.2	199.3	806.6	160769	5.1
Methane	111.5	511.1	422.6	215933	6.8
Helium	4.2	20.8	125.0	2604	0.1

# Two examples

## Large LHe storage @ CERN

- Volume:  $4 \times 120 \text{ m}^3$
- Boil off:  $\sim 0.20 \text{ g/s}$  or  $0.12\%/ \text{day}$

## Conversion to LH<sub>2</sub>

- $0.01\%/ \text{day}$  or  $0.01 \text{ g/s}$

Picture: CERN



## Small LHe storage @ CERN

- Volume:  $5 \text{ m}^3$
- Boil off:  $\sim 0.036 \text{ g/s}$  or  $0.5\%/ \text{day}$

## Conversion to LH<sub>2</sub>

- $0.04\%/ \text{day}$  or  $0.02 \text{ g/s}$



# Conclusions

- Due to high liquefaction costs, zero boil-off should be the goal for all liquid hydrogen storage tanks.
- Currently, most ground -based liquid hydrogen storage tanks have perlite+vacuum insulation without active shielding, resulting in boil-off rates of 1-5%/day.
- Boil-off could be reduced to 0.01-0.05%/day by using storage tanks based on LHe technology (MLI insulation and active shielding).
- In case of large storage tanks, boil-off gas could be re-liquefied to reach zero-boil off.





**Thanks for your attention**

